Characterisation by ¹H NMR spectroscopy of enzymically derived oligosaccharides from alkali-extractable wheat-flour arabinoxylan

Harry Gruppen ^a, Rainer A. Hoffmann ^b, Felix J.M. Kormelink ^a, Alphons G.J. Voragen ^a, Johannis P. Kamerling ^b and Johannes F.G. Vliegenthart ^b

(Received November 11th, 1991; accepted February 21st, 1992)

ABSTRACT

Oligosaccharides obtained from alkali-extractable wheat-flour arabinoxylans by digestion with endo-(1 \rightarrow 4)- β -D-xylanase from *Aspergillus awamori* were fractionated by size-exclusion chromatography on Bio-Gel P-2 followed by high-performance anion-exchange chromatography, and subjected to monosaccharide analysis and ¹H NMR spectroscopy. The results revealed (1 \rightarrow 4)-linked β -D-xylopyrano-oligosaccharides partly 3- and/or 2,3-substituted with single α -L-arabinofuranosyl groups. The structures of 12 such arabinoxylan oligosaccharides were established.

INTRODUCTION

Wheat-flour arabinoxylans consist^{1,2} of a linear backbone of $(1 \rightarrow 4)$ - β -D-xylopyranosyl residues, with mainly single α -L-arabinofuranosyl groups attached through positions 2 and 3, but little is known about the distribution of the side chains. Periodate-oxidation studies^{3,4} indicated the presence of clusters of 1–4 contiguous branched β -D-Xylp residues. However, a study of water-soluble rye arabinoxylans⁵ favours small isolated clusters of singly and doubly branched residues. Enzymic hydrolysis of the cell-wall material from bamboo and graminaceous plants yielded di-, tri-, and tetra-saccharides, some of which were feruloy-lated⁶⁻¹¹.

Detailed ¹H NMR studies of arabinoxylan oligosaccharides, derived from a warm-water-extractable arabinoxylan fraction by digestion with an Aspergillus

^a Department of Food Science, Agricultural University, P.O. Box 8129, 6700 EV Wageningen (Netherlands)
^b Bijvoet Center, Department of Bio-Organic Chemistry, Utrecht University, P.O. Box 80075, 3508 TB
Utrecht (Netherlands)

Correspondence to: Professor A.G.J. Voragen, Department of Food Science, Agricultural University, P.O. Box 8129, 6700 EV Wageningen, Netherlands.

endo- $(1 \rightarrow 4)$ - β -D-xylanase, have been reported ^{12,13}. The structures of alkali-extractable arabinoxylans have been less well characterised, due to the difficulties in their purification. We have described the isolation ¹⁴ of highly purified, water-unextractable, alkali-extractable arabinoxylans from wheat flour, which were further characterised ¹⁵ and fractionated by anion-exchange chromatography ¹⁶. We now report on the structures of oligosaccharides derived from these arabinoxylans by digestion with endo- $(1 \rightarrow 4)$ - β -D-xylanase I from Aspergillus awamori CMI 142717, the specificity of which is different from that of the endo- $(1 \rightarrow 4)$ - β -D-xylanase previously used ^{12,13}.

EXPERIMENTAL

Materials.—Wheat alkali-extractable arabinoxylan (BE1-U) was prepared from water-unextractable cell-wall material of the soft-wheat variety Arminda^{15,16}. Endo- $(1 \rightarrow 4)$ - β -D-xylanase I was purified¹⁷ from Aspergillus awamori CMI 142717.

Preparation of arabinoxylan oligosaccharides. —A solution of wheat alkali-extractable arabinoxylan (80 mg) in 50 mM sodium acetate buffer (80 mL, pH 5.0) was incubated with endo-(1 \rightarrow 4)-β-D-xylanase I (0.4 μg/mL) for 24 h at 30°. After inactivation of the enzyme (10 min, 100°), the solution was concentrated to 3 mL under reduced pressure, and applied to a column (100 \times 2.6 cm) of Bio-Gel P-2 (200–400 mesh, Bio-Rad) at 60° and eluted with distilled water (17 mL/h). Fractions (2.4 mL) were assayed for total neutral sugar content 18. Appropriate fractions were combined, designated 1–13, and each was concentrated to 1.5 mL under reduced pressure. The column was calibrated using a mixture of xylose, maltose, raffinose, stachyose, and Dextran T150 (Pharmacia). The elution volumes of these compounds corresponds with those of fractions 1–4, and 13 (void), respectively.

Fractions 3–10 were subjected to high-performance anion-exchange chromatography (HPAEC), using a Dionex Bio-LC GPM-II quaternary gradient module equipped with a Dionex CarboPac PA-1 column (250 × 9 mm). Samples (5 × 300 μ L) were injected using a Spectra Physics SP8780 autosampler equipped with a Tefzel rotor seal in a 7010 Rheodyne injector valve. Elution (5 mL/min) involved linear gradients of sodium acetate in 0.1 M NaOH 0 \rightarrow 150 mM during 10 min, then 150 \rightarrow 500 mM during 30 min at 20°. The solvents were degassed and stored under helium, using a Dionex EDM module. The eluate was monitored by using a Dionex PED detector in the pulsed-amperometric detection (PAD) mode. A reference Ag/AgCl electrode was used with a working Au electrode, with the following pulse potentials and durations: E₁ 0.1 V and 0.5 s, E₂ 0.6 V and 0.1 s, E₃ -0.6 V and 0.1 s. The eluate was neutralised with M acetic acid and the appropriate fractions (1.2 mL) were combined, desalted using columns (30 × 80 mm) of Dowex 50W-X8 (H⁺) and AG3 X4A (HO⁻) resins (Bio-Rad) in series, and concentrated under reduced pressure, and the residue was air-dried.

Monosaccharide analysis.—Samples (30–70 μ g) were hydrolysed by 2 M trifluoroacetic acid for 1 h at 121°, and the acid was evaporated in a stream of air at 40°. For Bio-Gel P-2 fractions, each monosaccharide mixture was reduced in 1.5 M ammonia (0.2 mL) containing 75 mg of NaBH₄/mL, and the products were converted into their alditol acetates¹⁹ and analysed²⁰. myo-Inositol was used as internal standard. For Dionex fractions, each hydrolysate was dissolved in water and analysed on a CarboPac PA-1 column (250 × 4 mm), eluted (1 mL/min) with 0.1 M NaOH, using PAD analysis (as described above).

FABMS.—Positive-ion FAB-mass spectra were recorded with a Jeol JMS AX 505 W spectrometer (Xe beam of 6 kV, acceleration potential of 3 kV) equipped with an HP9000 data system. Each sample was dispersed in a glycerol matrix, and the mass range was scanned at 10 s/scan with a mass resolution of 1500.

¹H NMR spectroscopy.—Samples were repeatedly treated with D₂O (99.9 atom% D, MSD Isotopes), finally using 99.96 atom% D at pD ≥ 7. Resolution-enhanced 600-MHz ¹H NMR spectra were recorded with a Bruker AM-600 spectrometer (SON-hf-NMR facility, Department of Biophysical Chemistry, Nijmegen University, Netherlands), operating at a probe temperature of 27°. Chemical shifts (δ) are expressed in ppm and were measured by reference to internal acetone (δ 2.225 in D₂O at 27°)²¹. Full details of the HOHAHA spin-lock experiments and ROESY spectroscopy have been reported¹².

RESULTS AND DISCUSSION

The fractionation of the arabinoxylan digest on Bio-Gel P-2 is shown in Fig. 1, and the monosaccharide composition and yield of each combined fraction are given in Table I. Of the 13 fractions, 11 (1-10 and 13) were observed as separate peaks, 13 was eluted in the void volume, and 1-10 accounted for 72% of the arabinoxylan. Fractions 1 and 2, consisting of xylose only (Table I), each showed one peak in HPAEC, and are suggested to contain xylose and xylobiose, respectively.

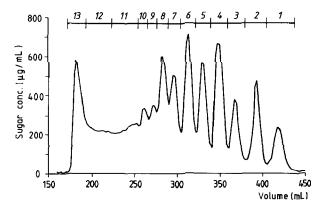


Fig. 1. Elution profile of the arabinoxylan digest on Bio-Gel P-2.

TABLE I

Data on the fractions obtained by chromatography on Bio-Gel P-2 and Dionex of the products formed by the enzymic degradation of wheat arabinoxylan

Bio-Gel P-2 fractions	ons													
	7	7	æ	4	S	9	7	∞	6	10	II	77	13	
Yield (%) a	4.4	6.9	9.9	10.6	8.4	11.1	7.7	8.9	4.1	3.9	8.8	8.3	10.9	
Ara (%) b	0	0	25	25	36	33	40	4 0	40	42	4	47	20	
Xyl (%)	100	100	75	75	61	<i>L</i> 9	09	09	09	28	26	53	20	
Dionex sub-fractions of $3-10$	ons of 3-	10												
	3.1	3.2	4.1	5.1	5.2	6.1	6.2	6.3	7.1	8.1	8.2	9.1	9.7	10.1
Recovery (%) c	31	89	86	8 4	44	09	27	12	82	53	28	64	27	47
Ara (%)	0	33	56	38	39	35	33	46	42	40	20	43	43	38
Xyl (%)	100	29	74	62	61	92	<i>L</i> 9	54	58	09	20	27	27	62
													1074	
Mol wt d	414	414	546	829	819	810	810	810	942	1074	1074	1206	1206	1338
	(3)	(3)	(4)	(5)	(5)	(9)	(9)	(9)	(7)	(8)	(8)	(6)	(8,9)	(10)

^a Percentage of total neutral sugars present in each fraction, based on spectrophotometric response of all sugars. ^b Sugar composition of the material in each peak. Percentage of total PAD response from each Bio-Gel P-2 fraction. Determined by positive-ion FABMS (number of pentose units in brackets).

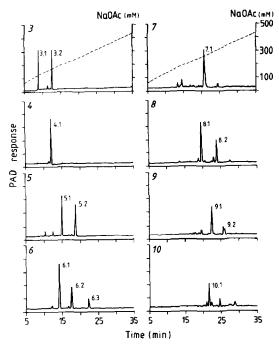


Fig. 2. Elution profile on HPAEC of Bio-Gel P-2 fractions 3-10 in Fig. 1.

Fractions 3-10 were subjected to HPAEC and the results are shown in Fig. 2. The monosaccharide compositions, molecular weights of the major components, and the recoveries are given in Table I. For fractions 4 and 7, no further fractionation was achieved, and fractions 4.1 and 7.1 represent 98 and 85%, respectively, of the total PAD responses. Fractions 3, 5, 8, and 9 each contained two major components, which represented > 90% of each total PAD response. Fraction 6 contained three oligosaccharides (6.1, 6.2, and 6.3) in the proportions 60, 27, and 12%, respectively. Fraction 10 contained one major component (47% of the total PAD response) and no attempt was made to isolate the minor components. The structures of the major oligosaccharides were elucidated by ¹H NMR spectroscopy.

Fraction 3.1.—The ¹H NMR spectrum of 3.1 matched exactly that of the reference compound Xyl_3 [β -D-Xylp-(1 \rightarrow 4)- β -D-Xylp-(1 \rightarrow 4)-D-Xylp]¹² and the chemical shifts of the H-1 resonances are summarised in Table II.

$$\begin{array}{ccc}
3 & 2 & 1 \\
\beta - Xyl p - (1 \rightarrow 4) - \beta - Xyl p - (1 \rightarrow 4) - Xyl p
\end{array}$$
3.1

Fraction 3.2.—The intensities of the H-1 signals for 3.2 (Fig. 3) indicated an arabinosylxylobiose structure with the Xyl p units β ($J_{1,2}$ 7–8 Hz) and the Ara f unit α ($J_{1,2} \sim 1.6$ Hz)^{6.22}. By comparison with published ¹H NMR data for the

5.271

	•	♦	\$ \$ \$	\$\\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\	\$ \$ \$
Residue b	3.1 °	6.1	8.1	9.1	10.1
α-Xyl p-1	5.184	5.184	5.184	5.183	5.184
β -Xyl p -1	4.584	4.584	4.584	4.584	4.584
β -Xyl p -2 α	4.475	4.465	4.465	4.464	4.465
β -Xyl p - 2_{β}	4.479	4.468	4.468	4.467	4.468
β -Xyl p -3	4.461				
β-Xyl p-3 ^{III}		4.639	4.637	4.628	4.639
β-Xyl <i>p</i> -4		4.437			4.443
β-Xyl p-4 ^{II}			4.480		
β -Xyl p -4 III				4.578	
β-Xyl p-5			4.435	4.428	
β -Xyl p - 5 ^{III}					4.628
β-Xyl p- 6					4.436
α -Ara f -A ^{2X3}		5.225	5.225	5.222	5.224
α -Ara f -A ^{3X3}		5.274	5.270	5.293	5.271
α -Ara f -A ^{2X4}				5.242	
α -Ara f -A ^{3X4}			5.402	5.281	
α -Ara f -A ^{2X5}					5.220

TABLE II

Chemical shifts ^a of the H-1 resonances of fractions 3.1, 6.1, 8.1, 9.1, and 10.1 from Table I

related feruloylated compound [5-O-(trans-feruloyl)- α -L-Ara f]-(1 \rightarrow 3)- β -D-Xyl p-(1 \rightarrow 4)-D-Xyl p)^{9,10}, the structure shown was assigned to 3.2. The ¹H NMR data are recorded in Table III.

$$\begin{array}{ccc}
2^{11} & 1 \\
\beta - Xyl p - (1 \rightarrow 4) - Xyl p \\
\end{array}$$

$$\begin{array}{ccc}
\alpha - Ara f - (1 \rightarrow 3) \\
A^{3X2}
\end{array}$$

3.2

Specific assignment of the α -Ara f H-5proR,5proS signals is based on their relative chemical shifts ($\delta_{5proR} > \delta_{5proS}$) supported by the $J_{4,5}$ values ($J_{4,5proR} < J_{4,5proS}$)²³. Owing to the presence of a reducing residue, there is an anomeric effect

^a Measured at 600 MHz on solutions in D₂O at 27° (internal acetone, δ 2.225). ^b The Xyl p residue in the reducing position is denoted 1, etc.; 2_{α} and 2_{β} mean that Xyl p-1 is α or β , Ara f-A^{2X3} means arabinofuranose linked to O-2 of Xyl p-3, etc., Xyl p-3¹ means Xyl p-3 branched at O-2, Xyl p-3¹¹ means Xyl p-3 branched at O-2, Xyl p-3¹¹ means Xyl p-3 branched at O-2, Xyl p; ϕ , α -Ara f: ϕ - ϕ , β -Xyl p-(1 → 4)-Xyl p; ϕ , α -Ara f-(1 → 2)- β -Xyl p; ϕ , α -Ara f-(1 → 3)- β -Xyl p.

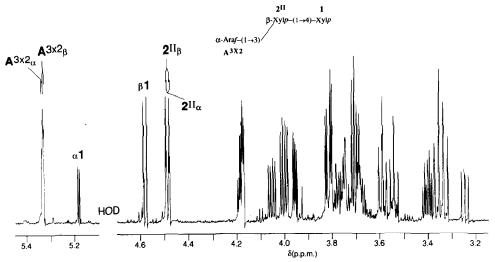


Fig. 3. Resolution-enhanced 600-MHz ¹H NMR spectrum of fraction 3.2. The numbers and letters in the spectrum refer to the corresponding residues in the structure.

with doubling of the H-1,2,3,4 signals of β -Xylp-2^{II} and the H-1 of (1 \rightarrow 3)-linked α -Ara f-A^{3X2}.

Fraction 4.1.—The intensities of the H-1 signals for 4.1 (Fig. 4) indicated an arabinosylxylotriose with the Xyl p units β ($J_{1,2}$ 7–8 Hz) and the Ara f unit α ($J_{1,2}$ ~1 Hz). Comparison of the ¹H NMR data (Table III) with those ¹² for AX-31 [β -Xyl p-4-(α -Ara f-A^{3X3}) β -Xyl p-3^{II}— β -Xyl p-2-Xyl p-1] shows that the β -Xyl p-4-(α -Ara f-A^{3X3}) β -Xyl p-3^{II} moiety is also part of 4.1, denoted β -Xyl p-3-(α -Ara f-A^{3X2}) β -Xyl p-2^{II}. However, the H-1 resonance (δ 4.509) of β -Xyl p-2^{II} of 4.1 has shifted slightly upfield (0.005 ppm), and the resonances of H-1,2,3 of β -Xyl p-2^{II} and H-1 of α -Ara f-A^{3X2} are double due to an anomeric effect, which accords with the linkage of the α -Ara f-(1 \rightarrow 3)[β -Xyl p-(1 \rightarrow 4)]- β -Xyl p group to a reducing xylose residue in 4.1.

4.1

Comparison of the ¹H NMR data (Table III) of **4.1** and **3.2** shows significant downfield shifts of the α -Ara f-A^{3X2} H-1,4,5 proS signals and significant upfield shifts of the H-2,3,5 proR signals, reflecting the effect of the change of the terminal α -Ara f-(1 \rightarrow 3)- β -Xyl p group to an α -Ara f-(1 \rightarrow 3)[β -Xyl p-(1 \rightarrow 4)]- β -Xyl p unit. The feruloylated compound β -D-Xyl p-(1 \rightarrow 4)-[5-O-(trans-feruloyl)- α -L-Ara f-(1 \rightarrow

$\begin{array}{cccccccccccccccccccccccccccccccccccc$	H-1	Cilcultai siiit				
		H-2	H-3	H-4	H-Seq/H-SproR	H-5ax /H-5proS
				3.73 — 3.82		
	1 4.584	3.250	3.545	3.781	4.055	3.377
			3.594	3.692)	000	2,70
			3.591	3.690 }	4.004	5.340
	A ^{3X2} 5.335 A ^{3X2} 5.332	4.175	3.959	4.185	3.817	3.705
				000		
		3.545	2 544	3.133.82-		3600
			3.344	3.//1	4.034	5.5/3
	2" 4.507 2" 4.509		3.749 3.746	3.829	4.124	3.400
			3.414	3.595	3.912	3.278
	A ^{3X2} 5 400)					
↓		4.158	3.903	4.272	3.796	3.714
, , ,	1 5.183	3.546		3.73—3.82		
>			3.546	3.771	4.050	3.376
			0	000		
		3.290 }	3.558	3.192	4.140	3.410
β -Xylp-3 III	3 ^{III} 4.596		3.689	3.724	4.023	3.344
α -Ara f -A ^{2X3}	A ^{2X3} 5.238		3.956	4.132	3.816	3.720
α -Ara f -A ^{3X3}		4.175	3.973	4.198	3.813	3.706
1-a X\[-a]	5.186	3.547		-3.73 - 3.82		
β -Xylp-1			3.546	3.781	4.055	3.377
>			245	t	70,	.04
β -Xyl p - 2^{II}_{B}		3.443	5.745	5.65/	4.120	3.401
β -Xyl p -3 $^{\text{ii}}$			3.581	3.663	3.948	3.312
α -Ara f -A $\frac{\alpha}{\alpha}$	A ^{3X2} 5.396 a ³ X2 5.391	4.159	3.902	4.270	3.796	3.715
a-Ara-f-A ^{3X3}		4.175	3.959	4.182	3.817	3.704

	3.377	3.399	3.366	3.269	3.719	3.715		3.373	3.455	•	3.309	3.728	3.745	3.704		3.376	3.417	3.431	3.307	3.720	3.720	3.704
	4.054	4.123	4.068	3.907	3.800	3.794		4.049	4.136	. !	4.056	3.816	3.839	3.811		4.051	4.143	4.146	3.956	3.817	3.796	3.815
-3.73 - 3.82 -	3.781	3.836	3.793	3.591	4.272	4.272	-3.73 - 3.82 -	3.770	3.830		3.68	4.307	4.15	4.196	-3.73 - 3.82 -	3.772	3.792	3.878	3.667	4.127	4.301	4.186
	3.545	3.740	3.735	3.409	3.914	3.900		3.544	3.764 \	3.701)	3.67	3.911	3.973	3.973		3.546	3.561	3.828	3.582	3.958	3.932	3.958
3 547	3.251	3.448 \	3.430	3.237	4.161	4.157	3.545	3.249	3.448	3.440	3.536	4.158	4.150	4.174	3.545	3.248	$\frac{3.298}{3.290}$	3.572	3.409	4.148	4.164	4.175
5 186	4.584	4.507	4.489	4.432	5.391 5.386	5.398	5.184	4.584	4.494	4.4%	4.563	$\frac{5.427}{5.422}$	5.244	5.244	5.184	4.584	4.467 4.467	4.639	4.467	5.225	5.271	5.331
2-Xvln-1	β -Xyl p -1	β -Xylp-2 $\frac{11}{\alpha}$	$\beta - Xy_1 p - z_{\beta}$ $\beta - Xy_1 p - 3^{11}$	β -Xyl p -4	α -Araf-A $_{\alpha}^{3X2}$ α -Araf-A $_{\beta}^{3X2}$	α -Ara f-A ^{3X3}	α -Xyl p-1	β -Xyl p-1	β -Xyl p -2 $\frac{11}{\alpha}$	β -Xyl p -2	β -Xyl p-3 ^{III}	α -Ara f -A $\frac{3}{3}$ α -Ara f -A $\frac{3}{8}$ α	α -Ara f -A ^{2X3}	α -Ara f -A ^{3X3}	α -Xv] p -1	β -Xyl p -1	β -Xyl p -2 α β -Xyl p -2 α	β -Xyl p-3 ^{III}	β -Xyl p-4 ^{II}	α -Ara f-A ^{2X3}	α -Ara f -A ^{3X3}	α -Ara f -A ^{3X4}
		> > 7					•		\$ 5	3												

TABLE III (continued)

Compound a	Residue "	Chemical shift b	shift ^b				
		H-1	H-2	H-3	H-4	H-5eq /H-5proR	H-5ax/H-5proS
	a-Xylp-1	5.184	3.544		3.73-3.8		
	β -Xyl p-1	4.584	3.247	3.545	3.771	4.051	3.376
۰ ۱	β -Xyl p -2 α	4.464	3.297	3 558	3 787	4 130	3.415
7.0	β -Xyl p -2 $_{\beta}$	4.466	3.288∫	טירייר	201.0	7.1.1	0.410
	β -Xyl p -3 ^{III}	4.626	3.565	3.85	3.86	4.139	3.491
	β -Xyl p -4 ^{III}	4.548	3.546	3.669	3.689	3.959	3.302
	α -Ara f -A ^{2X3}	5.221	4.152	3.957	4.125	3.816	3.720
	α -Ara f -A ^{3X3}	5.293	4.163	3.939	4.338	3.826	3.737
	α -Ara f -A ^{2X4}	5.254	4.149	3.973	4.16	3.838	3.737
	α-Ara f-A ^{3X4}	5.246	4.175	3.973	4.198	3.811	3.704

^a See Table II for the key. ^b In ppm relative to the signal of internal sodium 4,4-dimethyl-4-silapentane-1-sulfonate (using internal acetone at 8 2.225) in D₂O at 27°, acquired at 600 MHz.

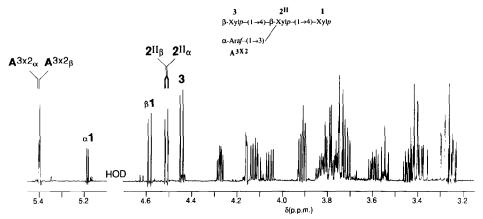


Fig. 4. Resolution-enhanced 600-MHz ¹H NMR spectrum of fraction 4.2. The numbers and letters in the spectrum refer to the corresponding residues in the structure.

3)]- β -D-Xylp-(1 \rightarrow 4)-D-Xylp has been described 10,24, and the reported 1H NMR assignments accorded those of comparable residues in 4.1.

Fraction 5.1.—The intensities of the H-1 signals for 5.1 (Fig. 5) indicated a single diarabinosylxylotriose with the Xylp units β ($J_{1,2}$ 7-8 Hz) and the Araf units α ($J_{1,2}$ 1-1.5 Hz). On the various H-1 tracks of the constituent monosaccharides in the 2D HOHAHA spectrum, the total scalar-coupled network for each residue was observed, and the data obtained are summarised in Table III. The observed ROEs along the H-1 tracks in the ROESY spectrum are compiled in Table IV. The ROEs between H-1 of β -Xylp-(n) and H-4,5eq of β -Xylp-(n-1),

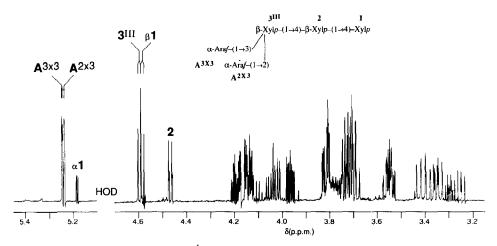


Fig. 5. Resolution-enhanced 600-MHz ¹H NMR spectrum of fraction 5.1. The numbers and letters in the spectrum refer to the corresponding residues in the structure.

TABLE IV

Cross-peaks observed at the H-1 tracks in the ROESY spectra of arabinoxylan oligosaccharides, measured with a mixing time of 200 ms

Compound	Residue	ROE effect
5.1	Xyl-2 H-1 Xyl-3 ^{III} H-1 Ara-A ^{2X3} H-1 Ara-A ^{3X3} H-1	Xyl-2 H-3,5 ax , Xyl-1 $β$ H-4,5 eq , Xyl-1 $α$ H-4,5 Xyl-3 III H-3,5 ax , Xyl-2 H-4,5 eq Ara- A^{2X3} H-2, Ara- A^{3X3} H-2, Xyl-3 III H-2 Ara- A^{3X3} H-2, Ara- A^{2X3} H-2, Xyl-3 III H-3
5.2	Xyl-2 ^{II} H-1 Xyl-3 ^{II} H-1 Ara-A ^{3X2} H-1 Ara-A ^{3X3} H-1	Xyl-2 ^{II} H-3,5 ax , Xyl-1 $β$ H-4 Xyl-3 ^{II} H-3,5 ax , Xyl-2 ^{II} H-4,5, eq Ara-A ^{3X2} H-2, Xyl-2 ^{II} H-3 Ara-A ^{3X3} H-2, Xyl-3 ^{II} H-3
6.3	Xyl- 2¹¹ H-1 Xyl- 3¹¹¹ H-1 Ara-A ^{3X2} H-1 Ara-A ^{2X3} H-1 Ara-A ^{3X3} H-1	Xyl- 2^{II} H-3,5 ax , Xyl- 1β H-4,5 eq , Xyl- 1α H-5 Xyl- 3^{III} H-3,5 ax , Xyl- 2^{II} H-4,5 eq Ara- 4^{3X2} H-2, Xyl- 2^{II} H-3 Ara- 4^{2X3} H-2, Ara- 4^{3X3} H-2 a , Xyl- 3^{III} H-2 Ara- 4^{3X3} H-2, Ara- 4^{2X3} H-2 a , Xyl- 3^{III} H-3
7.1	Xyl- 2 H-1 Xyl- 3^{III} H-1 Xyl- 4^{II} H-1 Ara-A ^{2X3} H-1 Ara-A ^{3X3} H-1 Ara-A ^{3X4} H-1	Xyl-2 H-3,5 ax , Xyl-1 $β$ H-4, Xyl-1 $α$ H-4,5 Xyl-3 ^{III} H-3(weak),5 ax , Xyl-2 H-4,5 eq Xyl-4 ^{II} H-3,5 ax , Xyl-3 ^{III} H-4,5 eq Ara-A ^{2X3} H-2, Ara-A ^{3X3} H-2, Xyl-3 ^{III} H-2 Ara-A ^{3X3} H-2, Ara-A ^{2X3} H-2, Xyl-3 ^{III} H-3 Ara-A ^{3X4} H-2(very weak), Xyl-4 ^{II} H-3
8.2	Xyl-2 H-1 Xyl-3 ^{III} H-1 Xyl-4 ^{III} H-1 Ara-A ^{2X3} H-1 Ara-A ^{3X3} H-1 Ara-A ^{2X4} H-1 Ara-A ^{3X4} H-1	Xyl-2 H-3(weak),5 α x, Xyl-1 β H-4,5 eq , Xyl-1 α H-5 Xyl-3 ^{III} H-3,5 α x, Xyl-2 H-4,5 eq Xyl-4 ^{III} H-3,5 α x, Xyl-3 ^{III} H-4,5 eq Ara-A ^{2X3} H-2, Ara-A ^{3X3} H-2, Xyl-3 ^{III} H-2 Ara-A ^{3X3} H-2(weak), Ara-A ^{2X3} H-2, Xyl-3 ^{III} H-3 Ara-A ^{2X4} H-2, Ara-A ^{3X4} H-2, Xyl-4 ^{III} H-2 Ara-A ^{3X4} H-2(weak), Ara-A ^{2X4} H-2, Xyl-4 ^{III} H-3

^a Due to the overlap of the H-1 resonances of A^{2X4} and A^{3X4}, the presence of these inter-residual ROE contacts could not be proved unambiguously, but were assumed to be present according to the structure 5.1.

together with the connectivities α -Ara f-A^{2X3}H-1, β -Xylp-3^{III} H-2 and α -Ara f-A^{3X3}H-1, β -Xylp-3^{III}H-3 established the sequence in 5.1.

$$\beta^{\text{III}} \qquad \qquad 2 \qquad 1 \\ \beta - \text{Xyl } p - (1 \rightarrow 4) - \beta - \text{Xyl } p - (1 \rightarrow 4) - \text{Xyl } p$$

$$\alpha - \text{Ara } f - (1 \rightarrow 3) \qquad \qquad A^{3X3} \qquad \alpha - \text{Ara } f - (1 \rightarrow 2)$$

$$A^{2X3} \qquad \qquad A^{2X3}$$

5.1

Comparison of the ¹H NMR data for 5.1 with those ¹² of AX-33 [β -Xylp-4-(α -Ara f-A^{2X3})(α -Ara f-A^{3X3}) β -Xylp-3^{III}- β -Xylp-2-Xylp-1] shows significant upfield shifts of all the signals of β -Xylp-3^{III} in 5.1 and there are differences in chemical

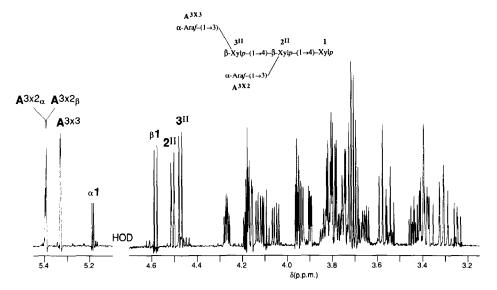


Fig. 6. Resolution-enhanced 600-MHz ¹H NMR spectrum of fraction 5.2. The numbers and letters in the spectrum refer to the corresponding residues in the structure.

shifts of the signals of the α -Ara f residues. The signals of α -Ara f-A^{3X3} are shifted in the same manner as found for α -Ara f-A^{3X2} on going from 4.1 to 3.2. For α -Ara f-A^{2X3}, only the H-1 resonance shows a small downfield shift (0.014 ppm), as compared to the same residue in AX-33. The inter-residual connectivities A^{2X3}-H-1,A^{3X3}-H-2 and A^{3X3}-H-1,A^{2X3}-H-2, found¹² for AX-33, were also present in the ROESY spectrum of 5.1, but with lower intensities.

Fraction 5.2.—The intensities of the H-1 signals for 5.2 (Fig. 6) indicated a diarabinosylxylotriose. The connectivities were determined as for 5.1 and the data are summarised in Tables III and IV. The ROEs between H-1 of β -Xylp-(n) and H-4,5eq of β -Xylp-(n - 1), together with the connectivities α -Araf-A^{3x2}H-1, β -Xylp-2^{II}H-3 and α -Araf-A^{3x3}H-1, β -Xylp-3^{II}H-3 established the sequence in 5.2.

A^{3X3}

$$\alpha$$
-Ara f -(1 \rightarrow 3)
$$\beta$$
-Xyl p -(1 \rightarrow 4)- β -Xyl p -(1 \rightarrow 4)-Xyl p

$$\alpha$$
-Ara f -(1 \rightarrow 3)
$$A^{3X2}$$

Comparison of the ¹H NMR data for 5.2 with those ¹³ of AX-41 [β -Xylp-5-(α -Ara f-A^{3X4}) β -Xylp-4^{II}-(α -Ara f-A^{3X3}) β -Xylp-3^{II}- β -Xylp-2-Xylp-1] shows large upfield shifts for all signals derived from β -Xylp-3^{II} relative to those of β -Xylp-4^{II} of AX-41, as expected on going from an internal to a terminal residue. The significant upfield shifts of the α -Ara f-A^{3X3} H-1,4,5proS signals and downfield

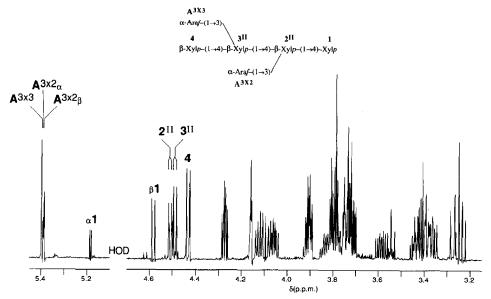


Fig. 7. Resolution-enhanced 600-MHz ¹H NMR spectrum of fraction 6.2. The numbers and letters in the spectrum refer to the corresponding residues in the structure.

shifts of the H-2,3,5 proR signals in 5.2, relative to the same signals of α -Ara f-A^{3X4} in AX-41, is noteworthy. Similar differences in chemical shifts were found for α -Ara f-A^{3X2} in 3.2 and 4.1 (see above). The ¹H NMR data of α -Ara f-A^{3X3} in 5.2 and of α -Ara f-A^{3X2} β in 3.2 are (almost) identical.

Fraction 6.1.—The intensities of the H-1 signals for 6.1 indicated a diarabinosylxylotetraose. The chemical shift data matched exactly those 12 of AX-33 and the chemical shifts of the H-1 resonances are summarised in Table II.

$$\beta - \text{Xyl } p - (1 \to 4) - \beta - \text{Xyl } p - (1 \to 4) - \beta - \text{Xyl } p - (1 \to 4) - \beta - \text{Xyl } p - (1 \to 4) - \text{Xyl } p$$

$$\alpha - \text{Ara } f - (1 \to 3)$$

$$A^{3X3} \quad \alpha - \text{Ara } f - (1 \to 2)$$

$$A^{2X3}$$

6.1

Compound 6.1 was also formed on enzymic treatment of barley flour 11.

Fraction 6.2.—The intensities of the H-1 signals for 6.2 (Fig. 7) indicated a diarabinosylxylotetraose. The connectivities were determined as for 5.1 and the data obtained are summarised in Table III. The ¹H NMR data that reflect¹³ the presence of the terminal β -Xylp-5-(α -Araf-A^{3X4}) β -Xylp-4^{II}-(α -Araf-A^{3X3}) β -Xylp-3^{II} moiety in AX-41 are also found for 6.2 and are assigned to β -Xylp-4-(α -Araf-A^{3X3}) β -Xylp-3^{II}-(α -Araf-A^{3X2}) β -Xylp-2^{II} and a reducing xylose residue. The finding of anomeric effects on the resonance of α -Araf-A^{3X2} H-1 and β -Xylp-2^{II} H-1,2 and a small upfield shift (δ 4.510) of the resonance of β -Xylp-2^{II} H-1,

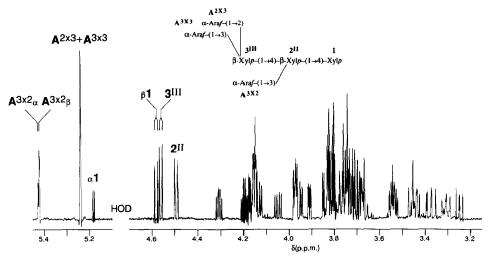


Fig. 8. Resolution-enhanced 600-MHz ¹H NMR spectrum of fraction 6.3. The numbers and letters in the spectrum refer to the corresponding residues in the structure.

compared to that (δ 4.514) of the β -Xylp-3^{II} H-1 resonance in **AX-41**, supports this conclusion (see also β -Xylp-2^{II} of **4.1** and **5.2**) and the structure assigned.

A^{3x3}

$$\alpha$$
-Ara f -(1 \rightarrow 3)

4
 β -Xyl p -(1 \rightarrow 4)- β -Xyl p -(1 \rightarrow 4)- β -Xyl p -(1 \rightarrow 4)-Xyl p
 α -Ara f -(1 \rightarrow 3)

A^{3x2}

6.2

Fraction 6.3.—The intensities of the H-1 signals for 6.3 (Fig. 8) indicated a triarabinosylxylotriose. The connectivities were established as for 5.1 and the data obtained are summarised in Tables III and IV. The ROEs between H-1 of β -Xylp-(n) and H-4,5eq of β -Xylp-(n - 1), together with the connectivities α -Ara f-A^{3X2} H-1, β -Xylp-2^{II} H-3, α -Ara f-A^{2X3} H-1, β -Xylp-3^{III} H-2 and α -Ara f-A^{3X3} H-1, β -Xylp-3^{III} H-3 established the sequence in 6.3.

$$A^{3X3}$$
 α -Ara f - $(1 \rightarrow 2)$
 α -Ara f - $(1 \rightarrow 3)$
 β -Xyl p - $(1 \rightarrow 4)$ - β -Xyl p - $(1 \rightarrow 4)$ -Xyl p
 α -Ara f - $(1 \rightarrow 3)$
 A^{3X2}

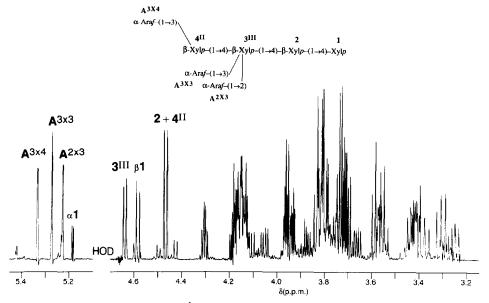


Fig. 9. Resolution-enhanced 600-MHz ¹H NMR spectrum of fraction 7.1. The numbers and letters in the spectrum refer to the corresponding residues in the structure.

Owing to overlap of the H-1 resonances of α -Ara f-A^{2X3} and α -Ara f-A^{3X3}, two sets of signals were observed along the HOHAHA H-1 track at δ 5.244. One set of signals match those of α -Ara f-A^{3X3} of **5.1** and were assigned to the corresponding residue in **6.3**. Owing to overlap of the α -Ara f-A^{2X3} H-2,4 signals, the chemical shift of the H-4 resonance could not be determined accurately. An anomeric effect was observed on the signals for β -Xylp-2^{II} H-1,2,3 and α -Ara f-A^{3X2} H-1, corroborating the structure assigned.

Fraction 7.1.—The intensities of the H-1 signals for 7.1 (Fig. 9) indicated a triarabinosylxylotetraose as the major compound. The connectivities were established as for 5.1 and the data obtained are summarised in Tables III and IV. The ROEs between H-1 of β -Xylp-(n) and H-4,5eq of β -Xylp-(n - 1), together with the connectivities α -Ara f-A^{2X3} H-1, β -Xylp-3^{III} H-2, α -Ara f-A^{3X3} H-1, β -Xylp-3^{III} H-3, and α -Ara f-A^{3X4} H-1, β -Xylp-4^{II} H-3 established the sequence in 7.1.

$$A^{3X4}$$
 α -Ara f -(1 \rightarrow 3)
$$A^{II} \qquad 3^{III} \qquad 2 \qquad 1$$
 β -Xyl p -(1 \rightarrow 4)- β -Xyl p -(1 \rightarrow 4)- β -Xyl p -(1 \rightarrow 4)-Xyl p

$$\alpha$$
-Ara f -(1 \rightarrow 3)
$$A^{3X3} \qquad \alpha$$
-Ara f -(1 \rightarrow 2)
$$A^{2X3}$$

Owing to the overlap of the H-1 signals of β -Xylp-2 and β -Xylp-4^{II}, two sets of signals were observed along the HOHAHA H-1 track at δ 4.467. Besides the ROEs mentioned above, the ROESY spectrum shows also the inter-residual connectivities A^{2X3} -H-1, A^{3X3} -H-2 and A^{3X3} -H-1, A^{2X3} -H-2, characteristic for the 2,3-substitution of a (1 \rightarrow 4)-linked β -Xylp residue by two terminal α -Araf groups^{12,13}. Comparison of the ¹H NMR data of 7.1 with those¹³ of AX-54c [β -Xylp-5-(α -Araf-A^{3X4}) β -Xylp-4^{II}-(α -Araf-A^{2X3})(α -Araf-A^{3X3}) β -Xylp-3^{III}- β -Xylp-2-Xylp-1] shows a good match for the resonances of β -Xylp-1,2,3^{III} and α -Araf-A^{2X3} and only small deviations for those of α -Araf-A^{3X3}, in particular H-3,5proR,5proS. The structural change in the non-reducing terminal part on going from 7.1 to AX-54c is reflected by the large upfield shifts for all signals of β -Xylp-4^{II}. The ¹H NMR data of α -Araf-A^{3X4} in 7.1 match those for α -Araf-A^{3X2} in 3.2 and α -Araf-A^{3X3} in 5.2, and support the structure assigned.

Fraction 8.1.—The intensities of the H-1 signals for 8.1 indicated a triarabinosylxylopentaose. The chemical shift data match exactly those¹³ of AX-54c, and the chemical shifts of the H-1 resonances are summarised in Table II.

8.1

Fraction 8.2.—The intensities of the H-1 signals for 8.2 (Fig. 10) indicated a tetra-arabinosylxylotetraose. The connectivities were established as for 5.1 and the data are summarised in Tables III and IV. The ROEs between H-1 of β -Xyl p-(n) and H-4,5eq of β -Xyl p-(n - 1), together with the connectivities α -Ara f-A^{2X3} H-1, β -Xyl p-3^{III} H-2, α -Ara f-A^{3X3} H-1, β -Xyl p-3^{III} H-3, α -Ara f-A^{2X4} H-1, β -Xyl p-4^{III} H-2 and α -Ara f-A^{3X4} H-1, β -Xyl p-4^{III} H-3 establish the sequence in 8.2.

$$A^{3X4} \quad \alpha - \text{Ara } f - (1 \to 2)$$

$$\alpha - \text{Ara } f - (1 \to 3)$$

$$\beta - \text{Xyl } p - (1 \to 4) - \beta - \text{Xyl } p - (1 \to 4) - \beta - \text{Xyl } p - (1 \to 4) - \text{Xyl } p$$

$$\alpha - \text{Ara } f - (1 \to 3)$$

$$A^{3X3} \quad \alpha - \text{Ara } f - (1 \to 2)$$

$$A^{2X3}$$

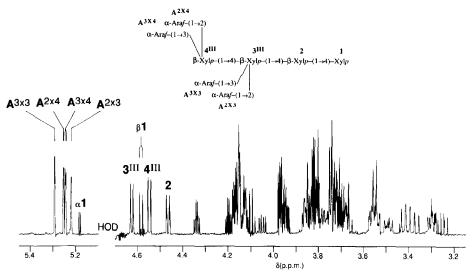


Fig. 10. Resolution-enhanced 600-MHz ¹H NMR spectrum of fraction 8.2. The numbers and letters in the spectrum refer to the corresponding residues in the structure.

The presence of the tetra-arabinosylxylobiose unit in **8.2** is reflected by a characteristic set of four α -Araf H-1 signals (Table III). The inter-residual α -Araf H-1,2 connectivities between α -Araf-A^{2X3} and α -Araf-A^{3X3} and between α -Araf-A^{2X4} and α -Araf-A^{3X4} support the presence of two 2,3-branched β -Xylp residues.

Fraction 9.1.—The intensities of the H-1 signals for 9.1 indicated a tetraarabinosylxylopentaose. The chemical shift data match exactly those¹³ of AX-59 and the chemical shifts of the H-1 resonances are summarised in Table II.

A^{3X4}
$$\alpha$$
-Ara f -(1 \rightarrow 2)
 α -Ara f -(1 \rightarrow 3) $\begin{vmatrix} \mathbf{3}^{\text{III}} & \mathbf{2} & \mathbf{1} \\ \mathbf{\beta}$ -Xyl p -(1 \rightarrow 4)- β -Xyl β -Ara β -(1 \rightarrow 2)
A^{3X3} α -Ara β -(1 \rightarrow 2)

The characteristic set of four α -Ara f H-1 and two β -Xyl p H-1 signals, which reflects the presence of an internal $(1 \rightarrow 4)$ -linked xylobiose unit composed of two 2,3-branched β -Xyl p with four α -Ara f residues, differs significantly from the set for 8.2.

Fraction 9.2.—The ¹H NMR spectrum of 9.2, which indicated the presence of at least two components, could not be unravelled due to its complexity. The

presence of a mixture was reflected by the HPAE-PAD chromatogram (Fig. 2) for fraction 9.

Fraction 10.1.—The intensities of the H-1 signals for 10.1 indicated a tetraarabinosylxylohexaose. The chemical shift data match exactly those¹³ of AX-57^{II} and the chemical shifts of the H-1 resonances are summarised in Table II.

Thus, of the arabinosylxylose oligosaccharides present in the enzymic digest of an alkali-extractable arabinoxylan from the soft-wheat variety Arminda, four (6.1, 8.1, 9.1, and 10.1) have been identified also in an enzymic digest of a warm-water-extractable arabinoxylan from Kadet wheat flour ^{12,13}. Comparison of the oligosaccharides obtained in this and the previous studies ^{12,13} showed remarkable differences. Since, for Arminda wheat flour, no major differences in branching pattern between water-extractable and alkali-extractable arabinoxylans were found ¹⁶, the differences in the oligosaccharides obtained are likely to reflect differences in enzyme specificity rather than differences in arabinoxylan structure. The patterns of action of various types of endo- $(1 \rightarrow 4)$ - β -D-xylanases on cereal arabinoxlans are under investigation.

ACKNOWLEDGMENTS

We thank C. Versluis for recording the FAB-mass spectra (Bijvoet Center, Department of Mass Spectrometry, Utrecht University). This investigation was supported by the Netherlands Program for Innovation Oriented Carbohydrate Research (IOP-k) with financial aid from the Ministry of Economic Affairs and the Ministry of Agriculture, Nature Management and Fisheries, the Netherlands Organization for Applied Scientific Research (TNO), and the Netherlands Foundation for Chemical Research (SON/NWO).

REFERENCES

- 1 G.B. Fincher and B.A. Stone, Adv. Cereal Sci. Technol., 8 (1986) 207-295.
- 2 R.A. Hoffmann, M. Roza, J. Maat, J.P. Kamerling, and J.F.G. Vliegenthart, *Carbohydr. Polym.*, 15 (1991) 415-430.
- 3 C.M. Ewald and A.S. Perlin, Can. J. Chem., 37 (1959) 1254-1259.
- 4 D.G. Medcalf and K.A. Gilles, Cereal Chem., 45 (1968) 550-556.
- 5 P. Aman and S. Bengtsson, Carbohydr. Polym., 15 (1991) 405-414.
- 6 M.M. Smith and R.D. Hartley, Carbohydr. Res., 118 (1983) 65-80.
- 7 F. Gubler, A.E. Ashford, A. Bacic, A.B. Blakeney, and B.A. Stone, *Aust. J. Plant Physiol.*, 12 (1985) 307–309.

- 8 H.R. Goldschmid and A.S. Perlin, Can. J. Chem., 41 (1963) 2272-2277.
- 9 I. Mueller-Harvey, R.D. Hartley, P.J. Harris, and E.H. Curzon, Carbohydr. Res., 148 (1986) 71-85.
- 10 T. Ishii and T. Hiroi, Carbohydr. Res., 196 (1990) 175-183.
- 11 K. Bock, J.Ø. Duus, B. Norman, and S. Pedersen, Carbohydr. Res., 211 (1991) 219-233.
- 12 R.A. Hoffmann, B.R. Leeflang, M.M.J. de Barse, J.P. Kamerling, and J.F.G. Vliegenthart, *Carbohydr. Res.*, 221 (1991) 63-81.
- 13 R.A. Hoffmann, T. Geijtenbeek, J.P. Kamerling, and J.F.G. Vliegenthart, *Carbohydr. Res.*, 223 (1992) 19-44.
- 14 H. Gruppen, R.J. Hamer, and A.G.J. Voragen, J. Cereal Sci., 13 (1991) 275-290.
- 15 H. Gruppen, R.J. Hamer, and A.G.J. Voragen, J. Cereal Sci., (1992) in press.
- 16 H. Gruppen, R.J. Hamer, and A.G.J. Voragen, J. Cereal Sci., (1992) in press.
- 17 F.J.M. Kormelink, M.J.F. Searle-vanLeeuwen, T.M. Wood, A.G.J. Voragen, and W. Pilnik, in C. Grassi, G. Gosse, and G. dosSantos (Eds.), *Biomass for Energy and Industry*; 5th E.C. Conference, Elsevier Applied Science, London, 1990, pp. 2.66-2.74.
- 18 M.T. Tollier and J.P. Robin, Ann. Technol. Agric., 28 (1979) 1-15.
- 19 H.N. Englyst and J.H. Cummings, Analyst, 109 (1984) 937-942.
- 20 H. Gruppen, J.P. Marseille, A.G.J. Voragen, R.J. Hamer, and W. Pilnik, J. Cereal Sci., 9 (1989) 247-260.
- 21 J.F.G. Vliegenthart, L. Dorland, and H. van Halbeek, Adv. Carbohydr. Chem. Biochem., 41 (1983) 209-374.
- 22 K. Mizutani, R. Kasai, M. Nakamura, O. Tanaka, and H. Matsuura, Carbohydr. Res., 185 (1989) 27-38.
- 23 G.D. Wu, A.S. Serianni, and R. Baker, J. Org. Chem., 48 (1983) 1750-1757.
- 24 A. Kato, J.-I. Azuma, and T. Koshijima, Agric. Biol. Chem., 51 (1987) 1691-1693.